

C2—C3	1.537 (6)	C8—C9	1.531 (5)
C2—C7	1.511 (4)	C8—C10	1.527 (6)
C2—C8	1.553 (5)	N1—H1a	0.89 (4)
C2—C11	1.502 (5)		
N3—N2—C1	118.5 (3)	C4—C5—C6	106.3 (4)
N2—N3—C7	116.8 (3)	C4—C5—C8	102.6 (4)
O1—C1—N1	122.9 (3)	C6—C5—C8	103.3 (3)
O1—C1—N2	121.0 (3)	C5—C6—C7	101.5 (3)
N1—C1—N2	116.1 (3)	N3—C7—C2	123.1 (3)
C3—C2—C7	104.5 (3)	N3—C7—C6	129.5 (3)
C3—C2—C8	101.8 (3)	C2—C7—C6	107.3 (3)
C3—C2—C11	113.8 (3)	C2—C8—C5	93.8 (3)
C7—C2—C8	99.3 (3)	C2—C8—C9	114.1 (4)
C7—C2—C11	116.0 (3)	C2—C8—C10	112.7 (3)
C8—C2—C11	119.2 (3)	C5—C8—C9	113.8 (3)
C2—C3—C4	103.8 (4)	C5—C8—C10	114.4 (4)
C3—C4—C5	103.1 (4)	C9—C8—C10	107.8 (4)

D—H...A	D—H	H...A	D...A	D—H...A
N1—H1b...O1 ⁱⁱⁱ	0.90 (4)	2.05 (4)	2.933 (4)	167 (4)
N2—H2...O1 ⁱ	0.83 (4)	2.31 (4)	3.137 (4)	177 (4)

Symmetry codes: (i) $\frac{1}{2} - x, -y, z - \frac{1}{2}$; (ii) $x, y, z - 1$; (iii) $\frac{1}{2} - x, -y, z + \frac{1}{2}$.

The scan rate was 6° min^{-1} (in ω). Weak reflections [$I < 10\sigma(I)$] were rescanned (maximum of three rescans) and the counts accumulated to improve accuracy. The positional parameters of atoms H1a, H1b and H2 were refined; all other H atoms were placed at assumed positions (C,N—H = 0.95 Å, $U = 1.2 \times U_{\text{eq}}$ of associated C atom) and fixed. Data Collection *MSC/AFC Diffractometer Control Software* (Molecular Structure Corporation, 1988). Cell refinement: *MSC/AFC Diffractometer Control Software*. Data reduction: *TEXSAN PROCESS* (Molecular Structure Corporation, 1985). Program(s) used to solve structure: *SHELXS86* (Sheldrick, 1985). Program(s) used to refine structure: *TEXSAN LS*. Molecular graphics: *TEXSAN, ORTEP* (Johnson, 1965). Software used to prepare material for publication: *TEXSAN FINISH*.

Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 71746 (15 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England. [CIF reference: CR1088]

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3-β-D-Ribofuranosyl-6,7-dihydro-9H-thiazolo[3,2-a]purin-9-one Hydrate

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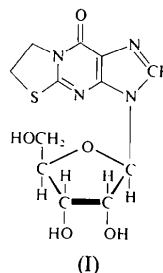
(Received 30 July 1993; accepted 25 October 1993)

Abstract

The title molecule, C₁₂H₁₄N₄O₅·S·H₂O (I), has a $\text{syn-}\chi_{\text{CN}}$ glycosyl torsion angle, which is stabilized by an intramolecular hydrogen bond between N3 of the tricyclic base and O5' of the ribose (in a C2'-endo pucker). [The purine base, including atoms S and O6, of the molecule is planar to within 0.043 (2) Å.] The tricyclic bases are stacked along *a* with an interplanar distance of 3.602 (3) Å.

Comment

As part of anti-tumour drug research, the structure of (I) has been determined in order to elucidate the sugar pucker and glycosyl torsion conformations. Details of the synthetic work have been published elsewhere (Chern, Lin & Chen, 1990).



The crystallographic analysis shows that the ribose ring adopts a $C2'$ -endo pucker conformation and that the intra-annular torsion angles starting from $C1'$ and proceeding around the $C1'-C2'-C3'-C4'-O4'$ ring are $-32.9, 18.1, 6.0, -27.7$ and 37.8 (3) $^\circ$, respectively. Interestingly, the *syn*-glycosyl conformation is stabilized by an intramolecular hydrogen bond [2.18 (4) Å] between N3 of the tricyclic base and O5' of the ribose. While the 'gauche effect' (Wolfe, 1972) suggested that ribose sugar favours a $C3'$ -endo pucker conformation, our results clearly indicate that other puckers are equally likely. Three hydrogen bonds from water and one between two molecules stabilize the lattice packing.

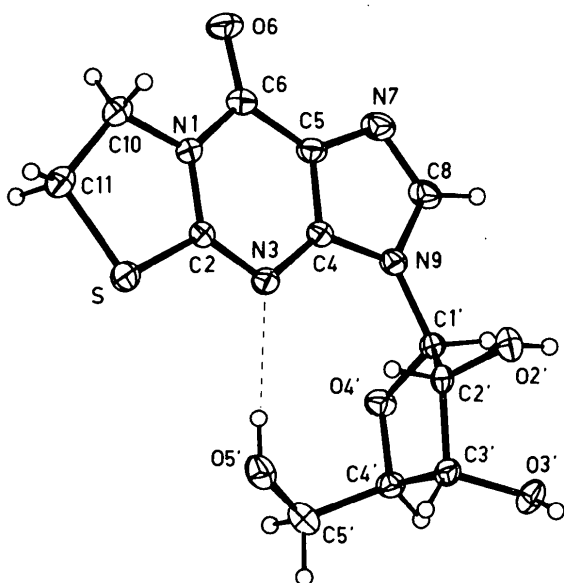


Fig. 1. Structure of (I) showing 50% probability displacement ellipsoids for the non-H atoms; H atoms are drawn as small spheres of arbitrary radii.

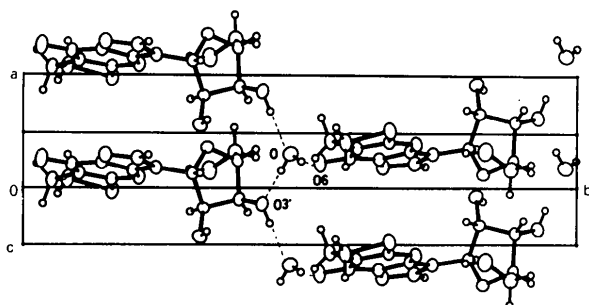


Fig. 2. Lattice packing of (I); four hydrogen bonds are drawn as dashed lines.

Experimental

Crystal data

$C_{12}H_{14}N_4O_5S \cdot H_2O$
 $M_r = 344.34$

Cu $K\alpha$ radiation
 $\lambda = 1.54178 \text{ \AA}$

Monoclinic

$P2_1$
 $a = 4.9635$ (4) Å
 $b = 21.935$ (2) Å
 $c = 6.7924$ (5) Å
 $\beta = 101.096$ (6) $^\circ$
 $V = 725.68$ (10) Å 3
 $Z = 2$
 $D_x = 1.576$ Mg m $^{-3}$

Data collection

Rigaku AFC-5R/RU300 diffractometer
 $\omega/2\theta$ scans [width (1.0 + 0.14tan θ) $^\circ$ in ω ; rate 4 $^\circ$ min $^{-1}$]
Absorption correction: empirical
 $T_{\min} = 0.92, T_{\max} = 0.95$
1253 measured reflections
1114 independent reflections

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.0222$
 $wR(F^2) = 0.0597$
 $S = 1.078$
1114 reflections
273 parameters
All H-atom parameters refined
Calculated weights
 $w = 1/[\sigma^2(F_o^2) + (0.0459P)^2 + 0.1009P]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.059$

Cell parameters from 24 reflections

$\theta = 25-40^\circ$
 $\mu = 2.363$ mm $^{-1}$
 $T = 293$ (2) K
Plate
0.4 × 0.35 × 0.2 mm
Colorless

1102 observed reflections
 $[I > 2\sigma(I)]$
 $R_{\text{int}} = 0.0132$
 $\theta_{\max} = 60.01^\circ$
 $h = 0 \rightarrow 5$
 $k = 0 \rightarrow 24$
 $l = -7 \rightarrow 7$
3 standard reflections
frequency: 150 min
intensity variation: none

$\Delta\rho_{\max} = 0.169$ e Å $^{-3}$
 $\Delta\rho_{\min} = -0.149$ e Å $^{-3}$
Extinction correction:
 $F_c^* = kF_c[1 + (0.001\chi \times F_c^2\lambda^3/\sin 2\theta)]^{-1/4}$
Extinction coefficient:
 $\chi = 0.0081$ (10)
Atomic scattering factors from *International Tables for Crystallography* (1992), Vol. C, Tables 4.2.6.8 and 6.1.1.4)

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters (Å 2)

$$U_{\text{eq}} = (1/3)\sum_i \sum_j U_{ij} a_i^* a_j^* \mathbf{a}_i \cdot \mathbf{a}_j$$

	x	y	z	U_{eq}
S	0.85196 (15)	0.65872 (3)	0.23980 (10)	0.0389 (2)
N1	1.1254 (5)	0.59994 (10)	0.0108 (4)	0.0308 (5)
C2	1.0496 (5)	0.65745 (14)	0.0539 (4)	0.0291 (6)
N3	1.1093 (5)	0.70757 (11)	-0.0337 (4)	0.0317 (5)
C4	1.2524 (5)	0.69581 (13)	-0.1813 (4)	0.0284 (6)
C5	1.3471 (5)	0.63969 (13)	-0.2328 (4)	0.0321 (6)
C6	1.2819 (6)	0.58606 (14)	-0.1334 (4)	0.0345 (6)
O6	1.3423 (5)	0.53263 (10)	-0.1608 (3)	0.0487 (6)
N7	1.4933 (5)	0.64608 (12)	-0.3854 (4)	0.0393 (6)
C8	1.4826 (6)	0.70456 (15)	-0.4239 (4)	0.0357 (7)
N9	1.3378 (4)	0.73700 (11)	-0.3056 (3)	0.0305 (5)
C10	1.0104 (8)	0.55108 (15)	0.1183 (5)	0.0396 (7)
C11	0.9275 (8)	0.5788 (2)	0.2989 (5)	0.0452 (8)
C1'	1.2927 (5)	0.80261 (13)	-0.3169 (4)	0.0272 (6)
C2'	0.9935 (5)	0.82235 (13)	-0.3328 (4)	0.0262 (6)
O2'	0.8287 (4)	0.81575 (10)	-0.5246 (3)	0.0355 (5)
C3'	1.0289 (5)	0.88725 (13)	-0.2526 (4)	0.0296 (6)
O3'	1.0669 (5)	0.92870 (10)	-0.4063 (3)	0.0391 (5)
C4'	1.2956 (6)	0.88370 (13)	-0.0947 (4)	0.0314 (6)
O4'	1.4349 (4)	0.82882 (9)	-0.1376 (3)	0.0337 (5)
C5'	1.2592 (7)	0.8800 (2)	0.1203 (5)	0.0437 (8)
O5'	1.0797 (5)	0.83291 (13)	0.1536 (3)	0.0483 (6)
O	0.5568 (5)	0.97957 (12)	0.5118 (4)	0.0463 (6)

Table 2. Selected geometric parameters (Å, °)

S—C2	1.742 (3)	C8—N9	1.375 (4)
S—C11	1.821 (3)	N9—C1'	1.456 (4)
N1—C2	1.364 (4)	C10—C11	1.496 (5)
N1—C6	1.396 (4)	C1'—O4'	1.407 (3)
N1—C10	1.473 (4)	C1'—C2'	1.531 (3)
C2—N3	1.311 (4)	C2'—O2'	1.406 (3)
N3—C4	1.360 (4)	C2'—C3'	1.522 (4)
C4—N9	1.359 (4)	C3'—O3'	1.424 (3)
C4—C5	1.386 (4)	C3'—C4'	1.537 (4)
C5—N7	1.382 (4)	C4'—O4'	1.446 (4)
C5—C6	1.424 (4)	C4'—C5'	1.508 (4)
C6—O6	1.233 (4)	C5'—O5'	1.410 (4)
N7—C8	1.308 (4)		
C2—S—C11	91.36 (15)	C4—N9—C8	106.2 (2)
C2—N1—C6	124.5 (2)	C4—N9—C1'	128.8 (2)
C2—N1—C10	114.6 (2)	C8—N9—C1'	125.0 (2)
C6—N1—C10	120.7 (2)	N1—C10—C11	107.7 (3)
N3—C2—N1	125.6 (2)	C10—C11—S	106.7 (2)
N3—C2—S	121.6 (2)	O4'—C1'—N9	108.2 (2)
N1—C2—S	112.8 (2)	O4'—C1'—C2'	105.3 (2)
C2—N3—C4	111.8 (2)	N9—C1'—C2'	114.9 (2)
N9—C4—N3	127.0 (3)	O2'—C2'—C3'	115.7 (2)
N9—C4—C5	105.8 (2)	O2'—C2'—C1'	114.8 (2)
N3—C4—C5	127.2 (3)	C3'—C2'—C1'	101.3 (2)
N7—C5—C4	110.6 (2)	O3'—C3'—C2'	110.9 (2)
N7—C5—C6	129.7 (3)	O3'—C3'—C4'	109.0 (2)
C4—C5—C6	119.7 (2)	C2'—C3'—C4'	103.1 (2)
O6—C6—N1	119.8 (3)	O4'—C4'—C5'	107.6 (2)
O6—C6—C5	129.0 (3)	O4'—C4'—C3'	106.4 (2)
N1—C6—C5	111.1 (3)	C5'—C4'—C3'	115.5 (3)
C8—N7—C5	104.0 (2)	C1'—O4'—C4'	109.6 (2)
N7—C8—N9	113.3 (3)	O5'—C5'—C4'	113.1 (3)

Table 3. Contact distances (Å)

A	H	D	A...H	D...A
N3	H5O'	O5'	2.21 (6)	3.045 (4)
O ^j	H3O'	O3'	1.93 (4)	2.725 (4)
O6	H1	O	2.03 (4)	2.784 (3)
O3 ⁱⁱ	H2	O3	1.98 (6)	2.826 (4)
O5 ⁱ	H2O'	O2'	1.98 (4)	2.742 (3)

Symmetry codes: (i) $x, y, 1 - z$; (ii) $1 - x, y, 1 + z$.

Dispersion corrections and absorption coefficients were taken from *International Tables for Crystallography* (1992, Vol. C, Tables 6.1.1.4, 4.2.6.8 and 4.2.4.2). Since (I) crystallizes in a polar space group, polar-axis restraints were applied by the method of Flack & Schwarzenbach (1988) and the absolute structure of the crystal was established as described by Flack (1983), which is consistent with the known absolute configuration of β -D-ribofuranose. Data collection and cell refinement: Rigaku AFC-5 software. Data reduction: *NRCVAX* (Gabe, Lee & Le Page, 1985). Program used to solve structure: *SHELXS86* (Sheldrick, 1990). Program used to refine structure: *SHELXL* (Sheldrick, 1994). Molecular graphics: *ORTEP* (Johnson, 1971). Software used to prepare material for publication: *SHELXL*; *NRCVAX*.

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Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 71785 (8 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England. [CIF reference: OH1058]

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Acta Cryst. (1994). **C50**, 736-738

Absolute Stereostructure of 13,17-Epoxy-*alisol B* 23-Acetate Isolated from *Alisma orientale*

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Abstract

The absolute configuration of the epoxy group in 13 β ,17 β :-24(*R*),25-diepoxy-1 β -hydroxydammar-3-one 23(*S*)-acetate, C₃₂H₅₀O₆, m.p. 472-474 K, [α]_D+139.4° ($c = 0.96$, CHCl₃), isolated from *Alisma orientale* rhizomes, was established to be the β orientation by X-ray crystallographic analysis. The *A*, *B* and *C* rings have chair forms. An intermolecular hydrogen bond is observed between O2 and O6' with a distance of 3.148 (6) Å.

Comment

13,17-Epoxyalisol B 23-acetate (I) was isolated from *Alisma orientale* rhizomes along with 20 related triterpenes of the protostane type (Nakajima, Mikoshiba, Ida & Shoji, 1984; Ida, Satoh, Nakajima, Yamaguchi & Shoji, 1989).

